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4 **Enabling food security through use of local rocks and minerals**

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18 **Abstract**

19

20 In many developing countries, replacement of the nutrients needed to produce subsistence and

21 cash crops is a major challenge, because of cost and long/complex supply chains. Nutrient audits

22 show that major nutrients are being removed from soils faster than they are being replenished,

23 which is clearly unsustainable. The use of crushed silicate rocks as a source of plant nutrients

24 predates the use of the chemical fertilizers that have revolutionised global agriculture. Such highly

soluble fertilizers are not ideal for the deeply leached oxisols widespread in the global south, and are rapidly leached. In these soils, silica may also need to be added as nutrient. In these circumstances, crushed silicate rocks have great potential to maintain soil health and to support crop production. In Brazil crushed rock remineralizers have been developed, and Brazilian federal law allows these to be used for crop nutrition, with specifications clearly defined by appropriate regulation. This approach provides a model that enables developing countries elsewhere to exploit local geological sources, and reducing dependency on imported chemical fertilizers. It creates opportunities for employment producing crushed rock products for different crops and locally variable soils and conditions, and illustrates renewed academic and practical interest in so-called 'Development Minerals.'

I. Introduction

The ability of soils to produce the food needed to support a global population expected to exceed 9 billion by 2050 is fundamental to sustainable development (Keesstra et al., 2016). Every crop removed from soil removes nutrients derived from the geological minerals present within the soil, and these need to be replaced, either by returning composted crop residues, manures etc, or by adding artificial fertilizers (Castellanos-Navarrete et al, 2015). Without careful husbandry, soils lose their ability to produce crops, threatening livelihoods at all levels and, more widely, the biodiversity of natural and managed (Brussard et al., 2007) ecosystems.

The importance of soil in economic development has been highlighted by Jeffrey Sachs (Sachs, 2003, 2005) as a key factor alongside disease and lack of infrastructure that needs to be addressed to support a population. More recently, this has been emphasised by Keesstra et al (2016) who

50 analyse the role of soil science in addressing the Sustainable Development Goals, stressing the
51 need for an interdisciplinary approach. Minerals and materials that are mined, processed,
52 manufactured and used domestically, so called 'Development Minerals' (Franks et al., 2016) are
53 receiving greater recognition for their role in structural economic transformation and poverty
54 alleviation, with rock dust a potential pillar of this growing field (Hilson, 2016; Franks, 2017; Hinton
55 et al., 2017).

56

57 Soil nutrient audits consider the balance between inputs and outputs, on a regional scale
58 (Vitousek et al., 2009). Most consider N and P, with few addressing K. On a global scale, Sheldrick
59 et al. (2002) considered N, P and K, highlighting the far greater deficiency in K compared with N
60 and P. Despite their identification of the K deficit, few subsequent studies have addressed K,
61 concentrating instead on N and P. Cobo et al. (2010) analyse nutrient use in Africa, at a range of
62 scales. They confirm the conclusion that nutrient mining is a significant problem, whilst
63 highlighting inconsistencies between published studies. Römheld and Kirkby (2010) emphasise
64 the importance of K for crop (and animal and human) health, noting inconsistencies in current
65 knowledge and approaches.

66

67 The price of fertilizers varies with time (World Bank, 2018; Figure 1). Since 2000, prices peaked for
68 N and P fertilizers in 2008, when oil prices also peaked. The price of potash peaked later, in early
69 2009. In general, N fertilizer price is closely related to the oil price (Figure 2), reflecting the use of
70 methane as a raw material in the Haber-Bosch process (Smil, 2001) as well as the energy cost of
71 manufacture (Lægrid et al., 1999). The price of diammonium phosphate (DAP) also is highest
72 when oil is highest, and between 2000 and 2018 the price of KCl was greatest (Figure 2) when
73 Brent crude was cheapest. These price differences reflect the differences in production of NPK
74 fertilizers, each of which has a very different supply chain. Since 2013, the price of potash

fertilizers has remained constant for periods of several months, stepping up and down, to around \$200/tonne at present, about twice the price at the start of the century. Such periods of constant price indicate a control that is independent of the price of oil.

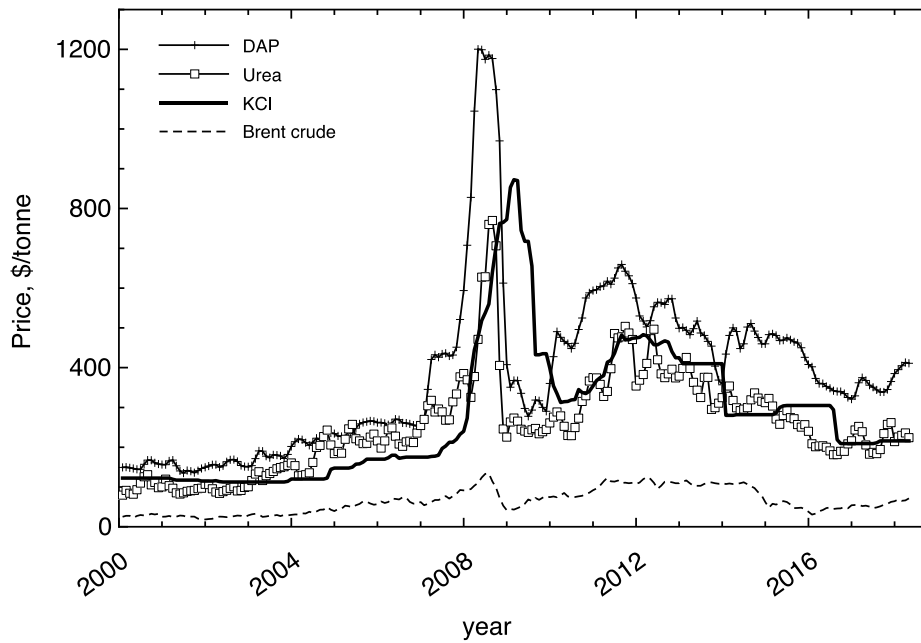
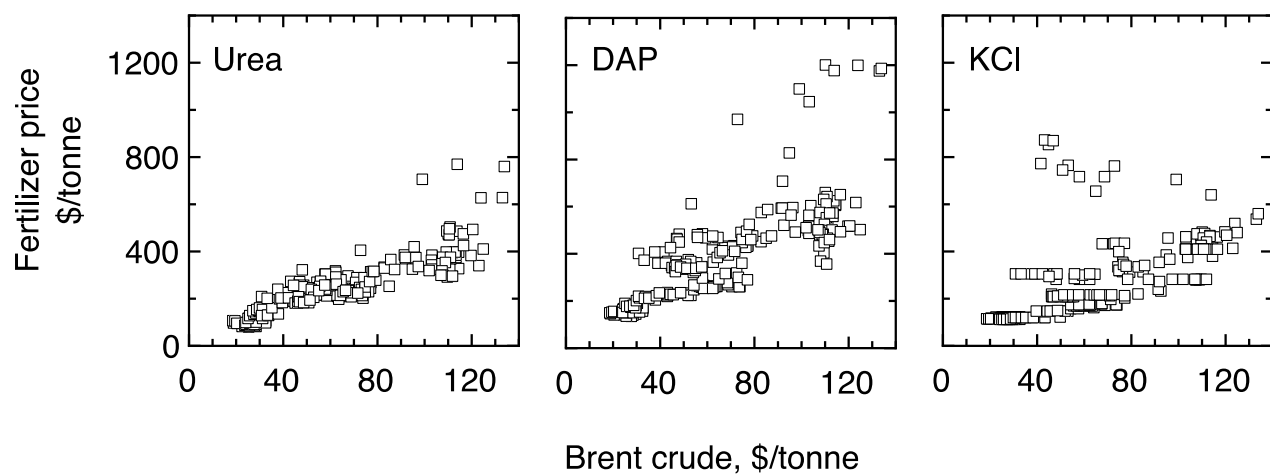


Figure 1. Variation in fertilizer prices, compared with oil (Brent crude) since 2000 (World Bank, 2018).



85 Figure 2. Variation in price of N, P and K fertilizers relative to price of Brent crude (2000-2018;
86 World Bank, 2018).

87

88

89 The ability of soils to continue to produce a harvest depends on their nutrient status (if other
90 factors are constant). The global use of fertilizers is recorded by the Food and Agriculture
91 Organization of the United Nations (FAO, 2016). When analysed on a regional basis, inequalities in
92 the use of fertilizers are readily apparent. Figure 3 shows per capita consumption, calculated using
93 the aggregate populations (United Nations Population Division, 2017) for different regions as
94 defined by the FAO. Approximate annual global consumption, per capita, for N is 13-16 kg, P (as
95 P_2O_5) is 6 kg and K (as K_2O) is 4-5 kg, rising over time for most regions. Per capita, North America
96 consistently consumes much more than the rest of the world, and South America more P and K
97 (neither of which are mined significantly in that region). Central Europe consumes more N than
98 other regions (apart from North America), but similar P and K. Regions which consistently show
99 least consumption are Africa (N,P,K), East Asia (N), West Asia (P,K) and South Asia (K). Thus FAO
100 figures show a major disparity in regional fertilizer use.

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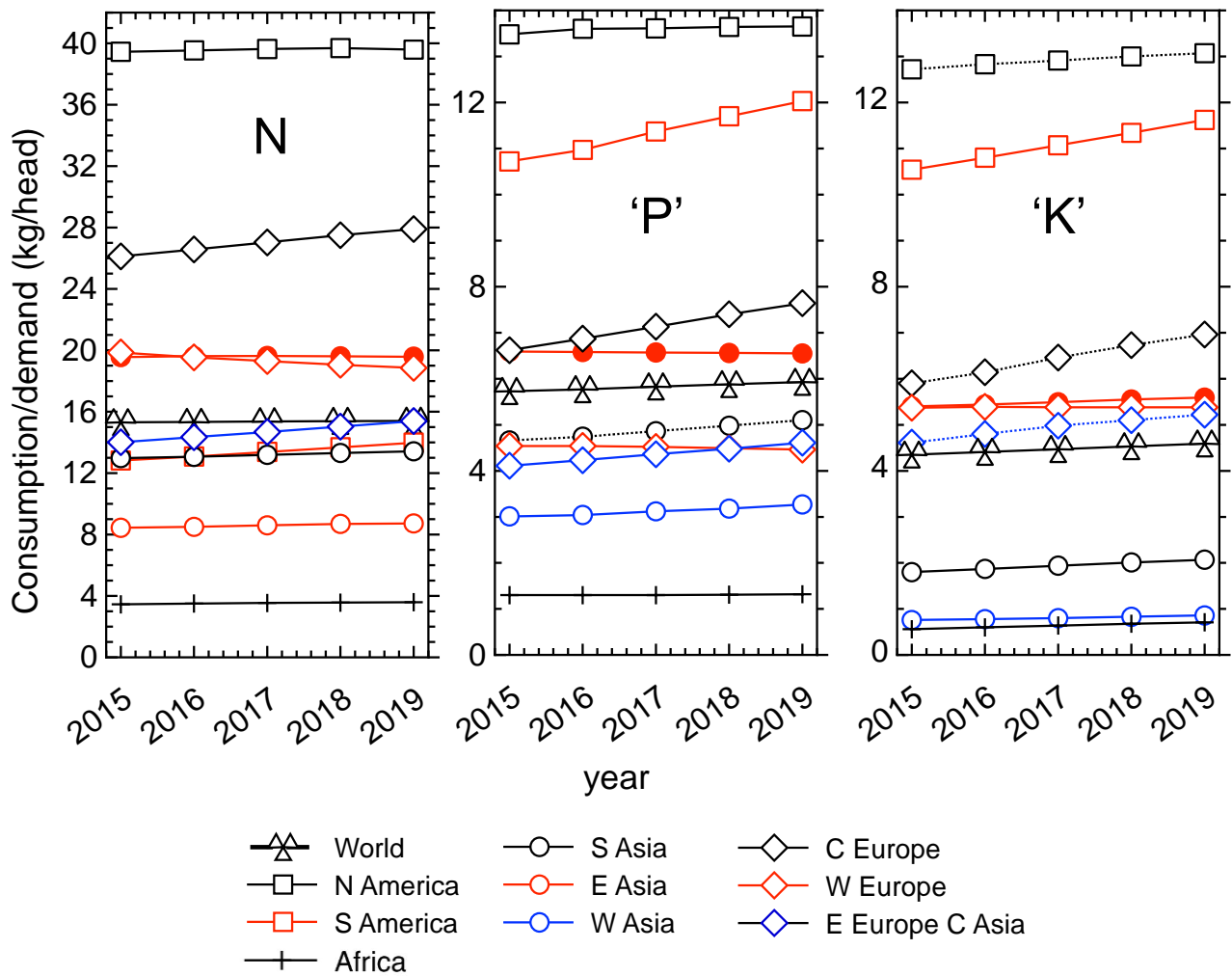
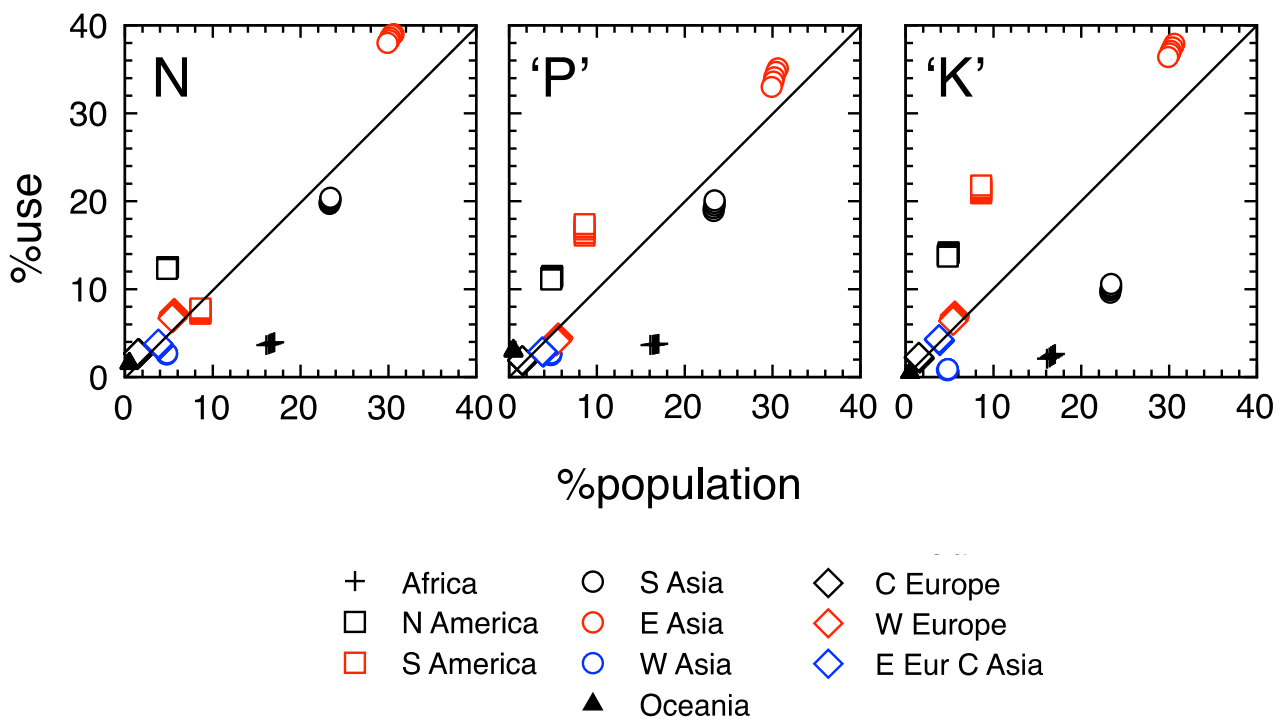


Figure 3. Regional use of fertilizers (data from FAO, 2016). 'P' denotes P_2O_5 ; 'K' denotes K_2O .

FAO figures have been recalculated to show proportions of consumption. In Figure 4, if the proportion of a region's population coincides with the proportion of fertilizer consumption, it plots on the 1:1 line diagonally crossing each graph. Consistently, North America and East Asia consume more than the proportion of population, and South America consumes more P and K. Consistently, Africa consumes less N, P and K than its population; South Asia and West Asia consume less K.

114 This approach to understanding demand for fertilizers represents a simplification of a complex
 115 situation. The use of conventional chemical fertilizers is part of a production model preferentially
 116 linked to agribusiness, which is directly related to the production of agricultural commodities. In
 117 South America (and especially in Brazil), fertilizer consumption is very high because the major
 118 countries of the region (Brazil and Argentina) are considered as agro-exporting countries,
 119 producing especially soybean, corn and wheat, all highly demanding of P and K. In Brazil, for
 120 example, according to data from ANDA (2016), about 65% of soluble fertilizers (NPK) are used for
 121 the production of three types of crop (soybean, corn and sugar cane).



124

125

126 Figure 4. Fertilizer consumption comparing population and use (data from FAO, 2016). 'P' denotes
 127 P_2O_5 ; 'K' denotes K_2O .

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129

130 Despite its limitations, the above analysis of price and use provides a global picture of inequality.
 131 In some regions, especially Africa, the use of fertilizers is far less than would be expected for a
 132 modern approach to agricultural production (Drechsel et al., 2001). Many factors contribute to
 133 this. Considering gender, Quisumbing and Pandolfelli (2010) state *“Although female heads of*
 134 *households uniformly apply less fertilizer than males, when farmer characteristics are controlled*
 135 *for in regression analysis, the critical factors that significantly limit fertilizer application are lack of*
 136 *access to credit and cash (Gladwin, 1992), not the sex of the farmer.”* Poor farmers in general, and
 137 particularly women, cannot afford fertilizers. In addition to poverty, which makes access to
 138 fertilizer resources difficult, women farmers are much more attached to the food and nutritional
 139 security of their families and, in many cases, consider that using the technological package
 140 (soluble fertilizers, agrochemicals and seeds) should be avoided. In the countries south of Ecuador,
 141 the number of women farmers who seek more agroecological alternatives (Siliprandi, 2014; 2018)
 142 is growing. Additionally, nutrient deficiencies, with consequences for human health, also reflect
 143 poor availability of trace nutrients in soils (eg Zn; Alloway, 2009).

144

145 In order to address these circumstances, alternative approaches to crop nutrition need to be
 146 considered (Horlings and Marsden, 2011). The primary goal is to enable farmers at the bottom of
 147 the economic ladder to produce crops first for subsistence and then for trade, enabling a greater
 148 number of people to participate in and benefit from modern agricultural methods. One approach
 149 is to use locally-obtained crushed rocks to supply a range of major and trace nutrients (van
 150 Straaten, 2009).

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152

153 II. Geochemical background to use of crushed rocks

154

155 The use of crushed rocks as sources of crop nutrients is long-standing, if not widespread. Driven
156 by necessity (Ciceri et al., 2015), Goldschmidt, the father of modern geochemistry, investigated
157 the use of nepheline syenite as a source of K for the US and UK at a time when Germany
158 dominated production and controlled international trade (Goldschmidt, 1922; Goldschmidt and
159 Johnsen, 1922). Leonardos et al (1987) proposed the use of silicate rocks as sources of crop
160 nutrients for lateritic soils, emphasising the differences between these and temperate soils, and
161 the implications for nutrient management. In lateritic soils, silicate minerals are demonstrably
162 unstable, as these soils are dominated by the presence of oxy-hydroxide minerals and clays
163 produced by silicate mineral weathering. In contrast, temperate soils contain a range of silicate
164 minerals predominantly produced by mechanical weathering of the parent rock. Modern chemical
165 fertilizers are designed for use in these, taking advantage of cation exchange as well as relatively
166 low levels of leaching. However, they are rapidly leached from lateritic tropical soils, which
167 generally have lower cation exchange capacities.

168

169 Plants require nutrients to be present in the soil solution, so that they can be taken up by roots.
170 The soil solution derives its mineral nutrients ultimately from two sources: the geological minerals
171 that are naturally present in the soil, or from artificial chemical fertilizers, which include salts that
172 are relatively soluble. The geological minerals that provide the major soil nutrients are dominated
173 by silicates, including micas, and these have a relatively low solubility. The mechanisms by which
174 nutrients are released to enter the soil solution and so become plant-available include weathering
175 (in which the original silicate mineral structure is destroyed) and cation exchange, in which the
176 structure is preserved, and cations are exchanged between the mineral and the soil solution.
177 Weathering of aluminium-bearing silicate minerals typically produces aluminous clay minerals,
178 reducing Al in solution to minimum values around neutral pH.

179

180 Silicate mineral weathering is well known but poorly quantified (White and Brantley, 1995), and is
181 essential for growth given the role of silica in plant metabolism (e.g. Keeping, 2017). In general,
182 the long-standing empirically observed weathering sequence (Goldich, 1938) indicates the relative
183 stabilities of the common rock-forming minerals. The thermodynamic basis for Goldich's
184 observations was provided by Curtis (1976); the weathering sequence closely corresponds to the
185 inherent thermodynamic stability of the minerals, and so is predictable. However, more
186 information is needed to predict the ability of a silicate mineral to release nutrients to the soil
187 solution, so that they become available to plants. The kinetics of the mineral dissolution reactions
188 control nutrient availability. These relate to the mineral's surface area and are normally expressed
189 as moles per metre squared per second and vary by several orders of magnitude (Palandri and
190 Kharaka, 2004). Table 1 gives examples of silicate mineral dissolution rates, expressed at standard
191 conditions of 25°C and pH 0 to allow comparison.

192
193 Importantly, the data shown in Table 1 have two messages. First, dissolution rate is not
194 necessarily greatest when the content of an element of interest, such as K, is highest. Nepheline
195 contains about 25% the K content of K-feldspar, but has a dissolution rate that is several orders of
196 magnitude greater. Anorthite has a Ca content smaller than wollastonite, and dissolves 60 times
197 more quickly. Secondly, the dependency on surface area shows the benefit of fine grinding (e.g.
198 Priyono and Gilkes, 2008), and (importantly) the need to know the surface area of mineral
199 treatments used in experimental work, taking this into account when comparing different studies.

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Mineral family	Mineral	Formula	Releases	Typical content	Dissolution rate mol.m ⁻² s ⁻¹	Relative dissolution rate
<i>Potassium silicates</i>						
K-feldspar	Orthoclase	KAlSi ₃ O ₈	K	14.1	-10.06	1
Feldspathoid	Leucite	KAlSi ₂ O ₆	K	17.4	-6.00	11,500
Feldspathoid	Nepheline	(Na,K)AlSiO ₄	K	4.2	-2.73	21,400,000
Mica	Muscovite	KAl ₃ Si ₃ O ₁₀ (OH) ₂	K	9.1	-11.85	0.016
Mica	Biotite	KMg ₃ AlSi ₃ O ₁₀ (OH) ₂	K, Mg	7.5, 3.6	-9.84	1.66
	Glauconite	K(Fe ³⁺ ,Al,Mg) ₂ (Si,Al) ₄ O ₁₀ (OH) ₂	K, Mg	7.5, 3.0	-4.80	182,000
<i>Other silicates</i>						
Plagioclase	Anorthite	CaAl ₂ Si ₂ O ₈	Ca	13.6	-3.50	3,630,000
Plagioclase	Albite	NaAlSi ₃ O ₈	Na	8.7	-10.16	0.794
Olivine	Forsterite	Mg ₂ SiO ₄	Mg	33.6	-6.85	1,620
Pyroxene	Wollastonite	CaSiO ₃	Ca	33.6	-5.37	49,000
Pyroxene	Diopside	CaMgSi ₂ O ₆	Ca, Mg	18.6	-6.36	5,010
Pyroxene	Enstatite	Mg ₂ Si ₂ O ₆	Mg	35.0	-9.02	11
Amphibole	Hornblende	Ca ₂ (Mg,Fe) ₄ Al[Si ₂ AlO ₂₂](OH) ₂	Ca,Mg	8.6, 7.8	-7.00	1,150
Tourmaline	Dravite	NaMg ₃ Al ₆ B ₃ Si ₆ O ₃₀ (OH)	Mg, B	7.8, 3.5	-6.50	3,630

205

206 Table 1 Silicate mineral dissolution rate constants (25°C, pH = 0; Palandri and Kharaka, 2004). For
 207 comparative purposes, relative dissolution rates express the dissolution rate of a mineral relative
 208 to that of K-feldspar.

209

210 Silicate rocks can be processed to yield concentrates of specific minerals, either as an intended
 211 product or a by-product (eg Bakken et al., 2000). This potentially adds cost, although it is a useful
 212 way of approaching a target of generating zero waste from a mine, delivering all mined material as
 213 a product. Without processing, use of crushed rock as a source of nutrients is an alternative route
 214 to nutrient supply. Given that crushed rocks typically contain more than one rock-forming

mineral, a bulk dissolution rate needs to be determined that takes into account (a) the minerals that are present and (b) their intergrowths and textural relationships. A general predictive model of element release from crushed rocks is a challenging task.

III. Experience in use of crushed rocks

A number of studies have investigated the ability of silicate minerals and rocks to supply nutrients for crop growth. In the UK and Sweden, some have shown little if any response by plants to treatment with crushed volcanic rock (e.g. Ramezani et al., 2012). Other studies have shown a positive response (e.g. Bakken et al., 2000). In detail, these contrasting results arise for mineralogical reasons. Bakken et al (2000) used a crushed rock that included nepheline; this has the highest dissolution rate of the silicate minerals listed in Table 1, and so might be expected to deliver K when exposed to a soil. Ramezani et al (2012) use a volcanic rockdust (pyroxene andesite), with a much coarser grain size distribution (Mohammed et al., 2014). This was applied to soils that in some cases contained the same minerals as the treatment – and, not surprisingly, no effect was observed. Mohammed et al (2014) took care to use natural soils that did not contain the same mineral as the treatment, and a response to the application of K silicate minerals was observed. Using artificial soils made from a high purity silica sand, Manning et al. (2017) showed that it was possible to demonstrate that plants are able to access K from a K-silicate mineral treatment, provided care was taken to ensure that the soil lacked other sources of K.

Other studies have investigated treatment with K-bearing silicate rocks (Manning, 2010; Harley and Gilkes, 2000). Many experiments in Europe and Australia have shown that application of crushed rock fails to provide K required for crop growth, in that statistically significant differences

240 are usually lacking. Manning (2010) reviewed 20 reports of 13 pot/9 field trials relating to K,
241 involving 16 crops and 7 different K-bearing silicate rock types. 13 of these reported benefits,
242 either as increased crop yield or increased K availability. In the absence of a consistent approach
243 from one study to another, it is difficult to take these results further to provide general
244 conclusions.

245

246 Overall, for soils in which silicate minerals already occur, like those dominating in temperate
247 zones, with a few exceptions, peer-reviewed experimental studies have yet to demonstrate
248 unambiguously that treatment with crushed silicate rocks gives an agronomic benefit. Yet in
249 practice farmers observe benefits and continue to use these materials. In contrast, there is an
250 increasing amount of information from studies carried out in Brazil, following early identification
251 of the potential of crushed silicate rocks in tropical soils (Ilchenko, 1955; Guimarães, 1955;
252 Leonardos et al., 1987; 2000). Most recently, Tavares et al. (2018) demonstrate the beneficial
253 effect of application of compost-phonolite mixtures for pasture. Guelfi-Silva et al. (2013)
254 compared a range of crushed rocks (milled, <3mm), including an alkaline volcanic breccia,
255 containing feldspathoids, zeolites and volcanic glass, an alkaline ultramafic rock containing olivine,
256 pyroxene, phlogopite, plagioclase and carbonate), two different biotite schists, containing biotite
257 and quartz, a phlogopite rock, containing phlogopite and serpentine, and a by-product of
258 manganese mining), with positive results for growth of lettuce in a Latosol. Santos et al (2016)
259 evaluate 'verdete' (<150 microns), a metamorphic schist containing glauconite (a mica) and K-
260 feldspar on growth of maize, grass and eucalyptus in pots, for Typic Hapludox soil. Without
261 pretreatment of the 'verdete', no benefit was seen. If calcined or acidified, agronomic benefits
262 were observed, as these chemical treatments increased K availability. Ramos et al (2017) consider
263 volcanic rock waste, assessing the availability of nutrients using leaching tests but without plant
264 growth experiments. Theodoro and Leonardos (2014) tested five types of rocks (fresh and

265 weathered basalt, kamaugite, carbonate schist and biotite gneiss) mixed (or not) with an organic
266 source. They verified that the availability of P, K, Ca and Mg increased in the soil, as compared to
267 the control plots, one year after the application of the remineralizers. They further showed that
268 remineralizers (mixed or not) with an organic source increased the pH and cation exchange
269 capacity values. The study also revealed that the production of five agricultural crops was better
270 following the second harvest, which may indicate that the solubility (and nutrient supply)
271 increases over time, with the interaction of the geological material in the soil and the organic acids
272 produced by roots. In other tropical systems, Anda et al (2009) demonstrate the value of crushed
273 basalt for improving the cultivation of cocoa on an oxisol (SE Asia; Rhodic Hapludox). In India,
274 Nishanth and Biswas (2008) showed the benefit of treatment based on a mixture of low grade
275 phosphate rock and muscovite for wheat production (Typic Haplustept soil), and Meena and
276 Biswas (2014) for microbial biomass and other soil parameters.

277

278 In Brazil, since the 1990s and particularly in the beginning of the 21st century, interest in using
279 crushed rocks to remineralize soils has increased, and has led to the formation of the Rochagem
280 movement. The need for alternative sources of crop nutrition arises from concerns about
281 environmental issues and rising fertilizer prices, giving Brazil's dependency on imports. The
282 major commodity crops (sugar cane, soybeans, maize, cotton, coffee beans, rice) are produced
283 from 76% of the country's agricultural land, representing 10% of rural properties. Production of
284 vegetables for sale to consumers (e.g. maize, beans, cassava, herbs and greens) is from 24% of the
285 country's land, with 90% of rural properties. This sector is dominated by small farmers, who find
286 access to fertilizers difficult because of cost as well as other factors. The Brazilian Federal
287 Government has taken steps to encourage small farmers to use more fertilizers, and to regulate
288 alternatives, to broaden the options available to this sector.

289

290 Access and use of conventional NPK formulations is strongly connected to a country's sovereignty.
291 Brazil is a typical example of this because, despite being the fourth largest fertilizer consumer in
292 the world (over the last 10 years it has imported about 70% of what it consumes), it is one of the
293 largest commodity producers globally (particularly soybeans; ANDA, 2016). This characteristic can
294 be interpreted in at least two ways. In the first, Brazil remains a strong agroexporter of
295 commodities, and in the second, its sector of greatest economic success is weakened as a result of
296 the need to import a large part of the input it consumes in order to continue producing. This
297 weakness opened up a space for discussing new technological paths, among which are the use of
298 soil remineralizers. These alternatives should be able to, simultaneously, have a positive change
299 on the low index of fertile tropical soil (which is highly weathered) and present results, in terms of
300 productivity, that are compatible with what farmers expect.

301

302

303 IV. The Rochagem movement in Brazil

304

305 In Brazil the Rochagem movement has been instrumental in making change that enables crushed
306 rocks to be used as remineralizers. It dates back to pioneering work carried out in the 1950s
307 (Ilchenko, 1955; Guimares, 1955). In the 1970s, Leonardos et al (1976) and Fyfe and Leonardos
308 (1978) proposed the use of volcanic rocks to recover soil fertility. This took place against the
309 background of the 'green revolution', that influenced agricultural policy in many countries,
310 especially in Brazil, involving the use of agrochemicals, improved seeds, and mechanization. At
311 that time, alternatives to this model were ignored by policy makers.

312

313 However, although it is unquestionable that this model of agricultural production was successful, in
314 terms of productivity, it has not fulfilled its principle promise, to end the world's hunger. On the

315 contrary, it has favoured the production of commodity crops in preference to food crops traded for
316 day-to-day consumption. Other associated problems have developed, with increased land
317 degradation and the continued exclusion of small, family-based, non-industrial agricultural units. In
318 these circumstances, research into the use of rock powders deserved to be considered. Towards
319 2000, research at the University of Brasilia demonstrated the value of ultra-potassic kamafugites for
320 the production of corn, sugar cane and manioc by farmers settled as part of the Program of Agrarian
321 Reform. This pioneering research demonstrated the need to form a network of researchers to test
322 the approach more widely in Brazil. At the same time, the Brazilian national agricultural research
323 organization, EMBRAPA, started research into the use of powdered rocks as an alternative to the
324 use of imported chemical fertilizers, because of their high price.

325

326 This coincidence of interests led to the organization in 2004 of the first international conference
327 “Rocks for Crops’ in Brasilia, with the participation of scientists and researchers from 5 continents,
328 representing Indonesia, Canada, Kenya, Japan and Portugal, as well as Brazilians already convinced
329 of the potential national importance of Rochagem. This conference considered the regulation of
330 rock powders in Brazil, and practical aspects of how rock powders would become part of Brazilian
331 agriculture. Three years later, in 2007, the second Rocks for Crops conference took place in Kenya,
332 with the participation of Brazilian researchers who presented new results from studies in Brazil.

333

334 An important outcome of these conferences, to improve consistency and to build on doctoral and
335 masters theses, was the need to establish in Brazil an Interinstitutional Working Group (Grupo de
336 Trabalho Interinstitucional - GTI) composed of researchers and technical experts from the
337 Government, involving the Ministries of Science and Technology, Mines and Energy, and
338 Agriculture, with universities, the Brazilian Geological Survey, Petrobras, the National Mining
339 Agency and EMBRAPA. This group organized the First Brazilian Congress on Rochagem in 2009

(Martins e Theodoro, 2009). At this event, the results of around 60 scientific studies were presented, by Brazilian scientists and international guests. The Second Brazilian Congress on Rochagem was held in 2013, with participants from 15 universities, six countries and representatives from the minerals industry. The conference proceedings contain around 70 papers (Theodoro, et al., 2013). One of the key themes discussed at this event was the establishment of parameters that could be used to define the permitted characteristics and guaranteed minimum specifications that would enable the use of remineralizers to be regulated. The Third Brazilian Congress on Rochagem was held in 2016, with around 80 scientific contributions (Bamberg et al., 2016).

Combined, these studies demonstrated positive results from the use of rock powders as remineralizers, including: (i) the costs of acquiring rock remineralizers are significantly less (up to 80%); (ii) a single application can be effective for up to four or five years; (iii) in remineralized soils, fertility levels have been increasing (particularly the levels of P, K, Ca and Mg) over the last five years; (iv) productivity is similar to or higher than conventional fertilization (it can yield up to 30% more return than for systems that use chemical inputs); (v) plant roots are better developed than those of plants which receive chemical fertilization, most probably due to the higher nutrient levels and reduced aluminum toxicity and pH correction; (vi) the level of soil moisture is higher in areas where remineralizers are applied, showing that they retain large amounts of water; (vii) plants show a higher amount of green mass, they are more abundant and have greater tillering; (viii) the plant's productive cycle was accelerated in some cases; (ix) there was no contamination or eutrophication of water sources because the rock dust has a gradual solubility, contrary to conventional fertilizers; and (x) it meets the standards of guarantee required of inputs used in organic agriculture, which as a sector has an average annual growth rate of 35% (Leonardos and Theodoro, 1999; Theodoro and Leonardos 2006; 2014; Souza et al. 2018; Melamhed, et al, 2009; Almeida et al, 2006)

365

366 Despite the many positive results presented in conferences, there was until recently no draft for a
367 formal protocol that could enable use, commercialization or regulation of ground rock soil
368 remineralizers since, due to their diverse characteristics, it was not possible to include them within
369 existing input categories (conditioners, fertilizers, etc.). Changing this gap in regulatory rules was a
370 fundamental factor in the viable use of remineralizers.

371

372 In order to change this situation, the interinstitutional working group developed a proposal which
373 was presented to the Brazilian legislature. The proposal was the culmination of discussions held in
374 national conferences, seminars and workshops on the proposal between the mineral sector and
375 the agriculture sector. This group came to the conclusion that the needs of the mineral sector to
376 develop new applications, in appropriate cases, for large amounts of residue resulting from
377 mineral extraction could be converted into a solution for agriculture as long as a few careful
378 measures were taken, such as: (i) a lack of contaminants in the crushed rock; (ii) having the main
379 macro and micronutrients in the rock minerals; and (iii) the availability of the source close to the
380 consumption area.

381

382 The Working Group suggested that a congressman present a Bill to the Federal Senate which
383 would include remineralizers in the Fertilizer Law (Law 6.894/1980). The congressmen understood
384 that this was an important issue that affected the sovereignty and development of the agriculture
385 sector in Brazil. Additionally, the international fertilizer market did not believe that this proposal
386 could have a strong impact on consumer demand. The few reactions to this were mainly local,
387 particularly because the potential consumers of these products were mostly family agriculturists.
388 The proposal to include rock dust as a category of agriculture input stated that remineralizers “*are*
389 *material of mineral origin whose size has been reduced and classified by mechanical processes*

390 *alone, its soil fertility indices altered through the addition of macro and micronutrients to the*
391 *plants which also helps to improve the physical and physical-chemical properties or the biological*
392 *activity of the soil”.*

393

394 The proposal was approved and passed relatively quickly in Brazilian National Congress (around 16
395 months). In October 2013 it became known as Law 12.890 (Brazil, 2013). Subsequently, a decree
396 was issued (Decree 8.384/2014) along with two Normative Instruction (INs) establishing minimum
397 requirements that remineralizers must meet in order to be recognized by Brazilian regulation. The
398 Instruction, IN 05 and 06 (Brazil, 2016), established the regulations for defining, classifying,
399 specifying and guaranteeing, tolerances, registering, packaging, labelling and marketing the
400 remineralizers used for agriculture.

401

402 This legal framework brought security and an increased interest on the part of Brazilian
403 agriculturists (including major soy producers) because it deals with an input which is available
404 locally/regionally, it is significantly cheaper and because the productivity is comparable to that of
405 regional averages. It sets out clear requirements for these materials to rebuild soil fertility and to
406 maintain crop production, in systems where conventional chemical fertilizers might be
407 inaccessible on the grounds of cost, or in organic systems where chemical products cannot be
408 used (Abbott and Manning, 2015). At present, a snapshot view of the on-line retail market (e.g.
409 <http://www.mfrural.com.br>) in Brazil shows that a number of different ‘pós de rochas’ products
410 (rock dust) are available at quoted prices from R\$60-350 (US\$15-90) per tonne.

411

412 Another positive point relating to the use of rock powders as remineralizers in Brazil concerns the
413 role played by the country’s minerals sector, which is a major exporter of metals, especially iron.

414 The extractive industries represent 4.3% of the entire domestic productivity of Brazil, and 16.9% of

415 industrial production (IBRAM, 2018). At present, according to data from the National Mining
416 Agency, there are 3354 mines active in Brazil, of which 5% (159) are large, producing >1million
417 tonnes per year, and 25% (837) are intermediate, producing 100 thousand-1million tonnes per
418 year. The great majority, (70%; 2358 mines) are small, producing less than 100000 tonnes per
419 year, often open pits that produce construction and civil engineering raw materials. These are
420 distributed widely across Brazil, providing great potential for the development of new products
421 targeted at agricultural markets. Many of these businesses produce materials that meet the
422 requirements of Lei 12.890/2013, including basalt, slate/shale and granites.

423

424 After the implementation of this law, many businesses showed an interest in registering with the
425 Ministry of Agriculture, Livestock and Supply to enable them to commercialise their products as
426 remineralizers. Obtaining registration has been a slow process, because the applicants must
427 demonstrate, through research and analysis, that their product is effective agronomically. Until
428 now, according to information from the Ministry, there have been around ten successful
429 registrations. It is hoped that this number will increase in the next few years, given the increasing
430 interest from conventional agriculture in the use of these inputs.

431

432 In addition, the National Policy of Organic Production (Política Nacional de Produção Orgânica;
433 PNAPO; Brazil, 2003) allows the use of rock powders in organic production systems, by means of
434 Instrução Normativa Nº 46, de 6 de outubro de 2011 (Anexo V; Brazil 2011). Considering that
435 demand for organic products is increasing by 30% annually, it is expected that use of appropriately
436 regulated rock powders as remineralizers will increase in these systems.

437

438 At this time, it is not possible to estimate the number of agricultural producers who use
439 remineralizers in Brazil. There is no formal statistical indicator of their use, but it is evident that

440 demand is growing, as experience shows that application of these geological materials improves the
441 fertility of Brazil's tropical soils.

442

443

444 V. Implications for development

445

446 The scientific basis for the use of crushed silicate rocks as sources of crop nutrients is growing in
447 extent, and explains why different results have been reported by studies in different parts of the
448 world. The Brazilian experience, which has led to formal recognition by government of farmers'
449 desire to use these materials and the development of an appropriate regulatory framework, has
450 implications for developing countries elsewhere in the world, and also for the developed world.

451

452 Translation of what has been learnt in Brazil to Africa has already started. Theodoro et al (2012)
453 describe a collaboration between Angola, Cameroon and South Africa, which enables a South-
454 South research network to facilitate transfer of knowledge from Brazil and vice versa. More
455 widely, acceptance of rock dust for soil remineralization in Brazil provides an example of a
456 regulatory framework for other countries. Use of rock dust as a source of K, for example, is
457 reported in studies from a range of countries and soil types (Manning, 2010), but with variable
458 results. A re-evaluation of these studies in the light of the theoretical understanding of the
459 dissolution of silicate minerals in soils allows greater insight into how they might be used, and
460 improved design of field and pot experiments, to ensure that these genuinely reflect the role of
461 the treatment.

462

463 The focus on nepheline-bearing rocks as a source of K has immediate relevance for areas close to
464 the East African Rift System (EARS). This is characterised by the presence of nepheline syenites

465 and related rocks, extending from Malawi to Ethiopia. Countries along the EARS include some of
466 the poorest in the world, and so identification of nepheline syenite as an indigenous source of K
467 may well be beneficial. If a regulatory system like that of Brazil is adopted by other countries, to
468 ensure safe use of mined materials, farmers have a new option for maintaining soil fertility. The
469 experience of using nepheline syenites as a source of K, documented in peer-reviewed research
470 papers, is long-standing (Goldschmidt 1922; Goldschmidt and Johnsen, 1922) and is extended by
471 the modern results reported by Bakken et al (2000) for Norwegian nepheline syenite, and also for
472 Brazilian phonolite (Tavares et al., 2018; Theodoro et al., 2012; phonolite is a fine grained variety
473 of nepheline syenite).

474

475 In the global north, use of crushed rock (rockdust) is accepted for remineralization in organic and
476 some conventional production systems. This approach aligns with 'agroecology', the desire to
477 approach agricultural production in a holistic way, so that food production is not in conflict with
478 other important ecosystem services. The concept of geotherapy is articulated by several authors
479 in Goreau et al (2014), and the UK's 25 year plan 'A Green Future: Our 25 Year Plan to Improve the
480 Environment' (DEFRA, 2018) sets a direction of travel that emphasises the importance of an
481 integrated approach to farming and ecosystem management. This approach has much to learn
482 from Brazil and the global south, where where opportunities to access economic and industrial
483 resources have driven the need to use alternatives to chemical fertilizers.

484

485

486 VI. Conclusions

487

488 The need in Brazil to find alternative sources of crop nutrition that are available to small farmers,
489 who produce the majority of non-commodity crops, has led to the development of a clear

490 regulatory framework that enables crushed silicate rocks to be used in circumstances where
491 conventional fertilizers are inaccessible, on the grounds of costs or logistics. Given that many
492 silicate rocks contain the nutrients required for plant growth, and occur widely, Brazil's approach
493 is applicable in many developing countries, especially those with deeply leached tropical soils.
494 The mechanisms by which crushed rocks release nutrients depend on dissolution rates of their
495 constituent minerals, rather than the content of the nutrient of interest. Once this is considered,
496 experiments to determine their efficacy can be designed and interpreted consistently. Wider use
497 of crushed silicate rocks provides one route to development especially of the agricultural sector
498 that produces crops for local markets. This approach enables farmers who cannot afford
499 conventional fertilizers to have an alternative, and particularly supports female producers.
500 In addition, this technological route can increase the potential of agro-ecological agriculture,
501 which is in agreement with a search for a more sustainable world, where food security is a
502 determinant axis of development.

503

504

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506

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509

510

511 VIII. References

512

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Mineral family	Mineral	Formula	Releases	Typical content	Dissolution rate mol.m ⁻² s ⁻¹	Relative dissolution rate
<i>Potassium silicates</i>						
K-feldspar	Orthoclase	KAlSi ₃ O ₈	K	14.1	-10.06	1
Feldspathoid	Leucite	KAlSi ₂ O ₆	K	17.4	-6.00	11,500
Feldspathoid	Nepheline	(Na,K)AlSiO ₄	K	4.2	-2.73	21,400,000
Mica	Muscovite	KAl ₃ Si ₃ O ₁₀ (OH) ₂	K	9.1	-11.85	0.016
Mica	Biotite	KMg ₃ AlSi ₃ O ₁₀ (OH) ₂	K, Mg	7.5, 3.6	-9.84	1.66
	Glauconite	K(Fe ³⁺ ,Al,Mg) ₂ (Si,Al) ₄ O ₁₀ (OH) ₂	K, Mg	7.5, 3.0	-4.80	182,000
<i>Other silicates</i>						
Plagioclase	Anorthite	CaAl ₂ Si ₂ O ₈	Ca	13.6	-3.50	3,630,000
Plagioclase	Albite	NaAlSi ₃ O ₈	Na	8.7	-10.16	0.794
Olivine	Forsterite	Mg ₂ SiO ₄	Mg	33.6	-6.85	1,620
Pyroxene	Wollastonite	CaSiO ₃	Ca	33.6	-5.37	49,000
Pyroxene	Diopside	CaMgSi ₂ O ₆	Ca, Mg	18.6	-6.36	5,010
Pyroxene	Enstatite	Mg ₂ Si ₂ O ₆	Mg	35.0	-9.02	11
Amphibole	Hornblende	Ca ₂ (Mg,Fe) ₄ Al[Si ₂ AlO ₂₂](OH) ₂	Ca,Mg	8.6, 7.8	-7.00	1,150
Tourmaline	Dravite	NaMg ₃ Al ₆ B ₃ Si ₆ O ₃₀ (OH)	Mg, B	7.8, 3.5	-6.50	3,630

Table 1 Silicate mineral dissolution rate constants (25°C, pH = 0; Palandri and Kharaka, 2004). For comparative purposes, relative dissolution rates express the dissolution rate of a mineral relative to that of K-feldspar.